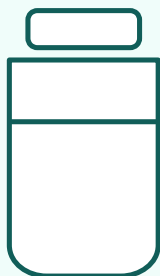


Alkaline Earth Metals

8th Grade

Group 2 — Reactive metals, but noticeably less so than group 1.



Reactive metals, but noticeably less so than group 1 — they hold their two outer electrons more tightly.

GROUP
Group 2

MEMBERS
Be, Mg, Ca

- Two electrons in the outer shell.
- Harder and denser than the alkali metals.
- Form ions with a +2 charge.
- Calcium is in bones; magnesium is in chlorophyll.
- Reactivity also increases going down.

LOOK FOR IT

Be · Mg · Ca · Sr · Ba · Ra

Elements in the same group share outer-electron counts, which is why they behave alike.